# Application and sensory evaluation of a hydroalcoholic extract of nasturtium (*Tropaeolum majus* L.) in a beverage from tropical fruits and vegetables

Aplicación y evaluación sensorial de un extracto hidroetanólico de capuchina (*Tropaeolum majus* L.) en una bebida de frutos y verduras tropicales

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## ABSTRACT

The nasturtium is diverse in bioactive compounds such as carotenoids, anthocyanins, and glucosinolates. Due to these attributes, it is important to apply them to novel foods, such as drinks. The objective of this study was to carry out a sensory evaluation for the viability of the addition of nasturtium extract to a beverage. Solvent extraction was performed at a solute-solvent ratio of 1:10 - 1:15 and ethanol purity of 45%, 70%, or 96%; antioxidant capacity was analyzed by the FRAP method and total phenolic compounds content by the Folin-Ciocalteu method. Additionally, three samples of the beverage that included nasturtium extract (extract content of 1000, 2500, or 5000 mg kg<sup>-1</sup>) were evaluated through a 5-point hedonic test and a preference test for the attributes of flavor, aroma, consistency, and the product as a whole. The samples with the highest solute-solvent ratio presented the highest antioxidant capacity, while the samples with ethanol purity of 45% and 70% obtained the highest content of total phenolic compounds. The sample with 1000 mg kg<sup>-1</sup> of the extract was the best qualified in all the sensory attributes evaluated.

**Key words:** phenolic compounds, fruit juices, hedonic scale, antioxidant capacity.

## RESUMEN

La capuchina es diversa en compuestos bioactivos como carotenoides, antocianinas y glucosinolatos. Debido a estos atributos, es importante aplicarlos a nuevos alimentos, como las bebidas. El objetivo del estudio fue realizar una evaluación sensorial de la viabilidad de la adición del extracto de capuchina en una bebida. La extracción con solventes se realizó con una relación soluto-solvente de 1:10 - 1:15 y pureza de etanol de 45%, 70% o 96%; la capacidad antioxidante se analizó por el método FRAP y el contenido de compuestos fenólicos totales por el método de Folin-Ciocalteu. Adicionalmente, se evaluaron tres muestras de bebida que incluía extracto de capuchina (contenido de extracto 1000, 2500 o 5000 mg kg<sup>-1</sup>) mediante una prueba hedónica de 5 puntos y una prueba de preferencia para los atributos de sabor, aroma, consistencia y el producto en general. Las muestras con mayor relación soluto-solvente presentaron mayor capacidad antioxidante, mientras que las muestras con pureza de etanol de 45% y 70% obtuvieron el mayor contenido de compuestos fenólicos totales. La muestra con 1000 mg kg<sup>-1</sup> de extracto fue la mejor calificada en todos los atributos evaluados.

**Palabras clave:** compuestos fenólicos, jugo de frutas, escala hedónica, actividad antioxidante.

### Introduction

Nasturtium (*Tropaeolum majus* L.) is a climbing plant of South American origin belonging to the Tropaeolaceae family (Brondani *et al.*, 2016), characterized by its round, peltate leaves and orange, yellow, and red flowers (Hegnauer, 1973). Its application is mainly ornamental, although it has also been used in traditional oral medicine for treatment of respiratory and skin infections (Alonso & Desmarchelier, 2015). These applications are attributed to bioactive compounds found in the plant, such as carotenoids, anthocyanins, glucosinolates and phenolic compounds (quercitrin, flavanols, gallic acid, caffeic acid, coumaric acid, chlorogenic acid) (Martínez-Navarrete *et al.*, 2008; Jakubczyk *et al.*, 2018; Demasi *et al.*, 2021). Due to the nasturtium attributes, the extraction of its bioactive compounds has been carried out through different methods, using solvents such as methanol (Navarro-González *et al.*, 2015; Demasi *et al.*, 2021) or acetone (Amiri, 2012), the Soxhlet method (Fernandes *et al.*, 2017) and new technologies such as ultrasound-assisted extraction (Jha Kumar & Sit, 2022); the extraction of phenolic and antioxidant compounds has been carried out using ABTS, DPPH methods (Garzón & Wrolstad, 2009; Arellano *et al.*, 2015), FRAP

Received for publication: August 29, 2022. Accepted for publication: April 20, 2023

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Doi: 10.15446/agron.colomb.v41n1.104490

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methods and HPLC (Garzón *et al.*, 2015; Navarro-González *et al.*, 2015). Despite this, there is still no application of these extracts or the plant in food products. Also, there are no antecedents or previous toxicological studies related to the intake of nasturtium. Therefore, the objective of this study was to do a sensory evaluation for the feasibility of adding nasturtium extract to a nectar-type drink, potentiating a novel beverage. It is important to clarify that this study is not trying to evaluate the interaction or antagonism between the components of the nectar drink and the extract.

# Materials and methods

### Plant material preparation

The nasturtium was harvested in the town of Cogua, in the department of Cundinamarca, Colombia and taken to the Pontificia Universidad Javeriana (Bogotá, Colombia), where it was separated into leaves, flowers, and calyx. Subsequently every part was freeze-dried in a FreeZone 4.5 LABCONO freeze-dryer at a temperature of -80°C and a pressure of 0.120 mbar and later ground in a KitchenAid grinder, to be packaged and stored at -20°C. For the extraction and evaluation methods, only the nasturtium leaves were used.

### Extraction

Solvent extraction water-ethanol was performed at room temperature, under constant stirring for 48 h, employing 5 g of the nasturtium leaves. Then, the solvent was filtered and concentrated in an IKA brand RV 10 digital V rotary evaporator, at a bath temperature of 50°C, rotation speed of 55 rpm and absolute vacuum of 100 mbar. Two extraction factors were evaluated: solute-solvent ratio (1:10 and 1:15 w/w), and ethanol purity in the solvent mixture (45%, 70% or 96%), performing six treatments in total and three replicates for every treatment, with the same extraction conditions. For every treatment, a new extraction was developed. A two-way analysis of variance and the interaction was performed for the statistical analysis, in addition to Tukey's multiple comparison test.

## Determination of antioxidant capacity and total phenolic compounds

The antioxidant capacity of the extract was determined by the FRAP spectrophotometric method following the method of Demasi *et al.* (2021), to include many compounds such as antioxidants, phenolic compounds, and vitamin C; the results were expressed in units of Trolox Equivalent Antioxidant Capacity - TEAC (µmol Trolox/g extract). The content of total phenolic compounds was analyzed by the Folin-Ciocalteu method (Navarro-González *et al.*, 2015), and the results were expressed in mg gallic acid equivalents / g extract. Each of the six treatments was analyzed in triplicate.

### Beverage development

A nectar-type drink of tropical fruits and vegetables was made: 25% w/w 'Valencia' orange juice (*Citrus sinensis* [L.] Osbeck), 23% w/w yacon pulp (*Smallanthus sonchifolius* [Pöpp. & Endl.] H. Rob.), 16% w/w 'Granny Smith' apple pulp (*Malus domestica* Borkh.), 10% w/w chayote pulp (*Sechium edule* [Jacq.] Swartz), 12% w/w of an infusion of mint (*Mentha spicata* L.) at 5% w/w and 11% w/w cucumber pulp (*Cucumis sativus* L.). A known concentration of the extract was added to this drink (see the sensory analysis method), together with additives such as xanthan gum (0.2%), maltodextrin (1%) and 1% of a commercial sweetener, Best4u (https://fitmarketbogota.co/products/ endulzante-natural-250gr-best4u), which is a mixture of erythritol and stevia.

## Sensory analysis

Three beverages were prepared with different extract content: 1000, 2500, or 5000 mg extract/kg beverage (the extract used in this experiment was obtained with ethanol with 70% of purity and relation solute-solvent 1:10; this extract had one of the highest total phenolic compounds compared to the other extracts evaluated). The beverages were analyzed through an affective test (hedonic scale of 5 points: 1: I dislike it very much, 2: I dislike it a little, 3; I neither like nor dislike it, 4: I like it a little, 5: I like it a lot) with 60 untrained consumers, evaluating three attributes (aroma, flavor, consistency) and the product in general. Each consumer was given 20 ml of each sample in glasses coded with 3 random digits, at a temperature of 10°C. In addition, a preference test was conducted, in which the evaluator was asked to place the sample code in front of the phrase that most identified it: "the one that I like the most - the one that follows me in taste - the one that I like the least."

For statistical analysis, an analysis of variance (ANOVA) was performed. When the results did not present normality, a non-parametric Kruskal-Wallis analysis was performed, with a confidence level of 95%.

# **Results and discussion**

#### Antioxidant capacity and total phenolic compounds

Table 1 presents the six experimental samples and their antioxidant capacity and total phenolic compounds.

The experiments obtained values higher than those reported by others, with 12.95 mg gallic acid equivalent/g for the content of phenolic compounds reported by Navarro-González *et al.* (2015) and 9.51 TEAC and 14.2 TEAC for the antioxidant capacity reported by Navarro-González *et al.* (2015) and Demasi *et al.* (2021), respectively. Both studies carried out extractions with methanol-water, a solvent mostly used for the extraction of bioactive compounds; however, due to its toxicity when ingested (Alcalá Pedrajas, 2002), it is necessary to review other solvents with equal or better yields to be used in edible products. Therefore, extraction with food-grade ethanol is a good replacement for methanol.

Extracts 4 and 5 had higher antioxidant capacity (Tab. 1), both with the highest solute-solvent ratio evaluated (1:15). Similar results were obtained by Yang and Li (2022), in which a ratio of 1:20 (the largest evaluated in the study) was the one with the highest antioxidant capacity in an extraction with 80% methanol solvent from the Angelica dahurica (Fisch.) Benth. & Hook plant. Sample 1, with a ratio of 1:10 and 96% ethanol, obtained the lowest value for total phenolic compounds, while samples 2 (70% ethanol - ratio 1:10) and 6 (45% ethanol - ratio 1:15) obtained the highest value for phenolic compounds. Both samples had solvent mixtures (ethanol-water); this indicates that for optimal extraction of phenolic compounds it is necessary to consider solvent mixtures, taking advantage of the affinities of the compounds with different solvents used. In the review by Jha Kumar and Sit (2022), different solvents were used to extract polyphenols from different plants, and the authors mentioned that, to extract compounds flavonols and phenolic acids, it is necessary to use ethanol in concentrations of 10%-90%. In addition, various studies obtained results similar to the present study, obtaining extracts with higher contents of total phenolic compounds with the ethanol concentration of up to 80% (Celant *et al.*, 2016; Medina-Medrano *et al.*, 2018; Martínez-Patiño *et al.*, 2019). Samples with the highest antioxidant capacity were not the ones with the highest phenolic content (samples 4 and 5), which would indicate the extraction of other compounds with antioxidant capacity found in nasturtium leaves, such as carotenoids and glucosinolates (Brondani *et al.*, 2016; Hernández-Rodríguez *et al.*, 2020; Demasi *et al.*, 2021).

#### Sensory evaluation

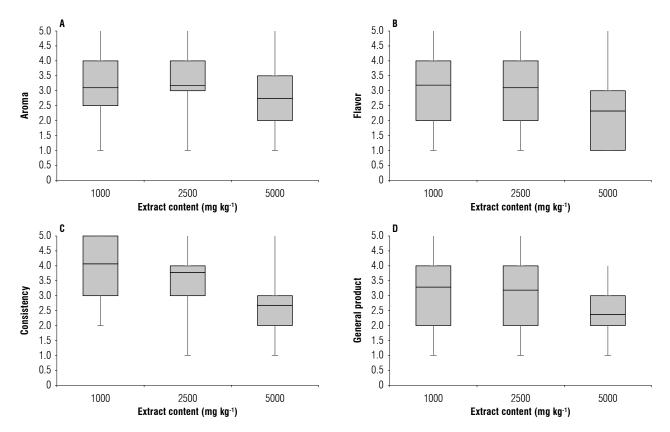
The mean aroma for all samples was similar, this attribute being a non-determining factor for the evaluators (Fig. 1). Sample 3 was the one with the lowest rating for all the attributes, its means being significantly different from the others (ANOVA  $P \le 0.05$ ), showing a clear rejection by consumers. On the other hand, the 1000 mg kg<sup>-1</sup> sample and the one with 2500 mg kg<sup>-1</sup> had similar means in the attributes of taste, consistency, and the product in general, but for the attribute of consistency, the difference is greater compared to the others (Fig. 1).

Regarding the preference test, the sample with 1000 mg kg<sup>-1</sup> was the highest qualified, followed by the sample with 2500 mg kg<sup>-1</sup> and, finally, the sample with 5000 mg kg<sup>-1</sup> (Fig. 2). Similar behavior was observed in the hedonic test, confirming that the drink with 1000 mg kg<sup>-1</sup> of the extract was the most preferred by the evaluators. Also, the consumers had no adverse effects due to the intake of the samples.

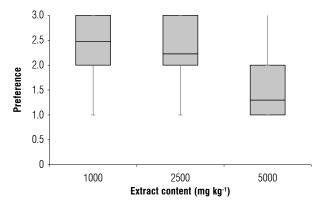
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Sample	Relation solute-solvent (w/w)	Purity of ethanol (% v/v)	Antioxidant capacity mean $\pm$ PSD <sup>1,3</sup>	Total phenolic compounds mean $\pm$ PSD <sup>2,3</sup>
1	1:10	96	19.86±0.35	2.60±0.50
2	1:10	70	24.49±0.35	39.58±0.50
3	1:10	45	$17.76 \pm 0.35$	16.91±0.50
4	1:15	96	$36.30 {\pm} 0.35$	21.87±0.50
5	1:15	70	32.21±0.35	33.75±0.50
6	1:15	45	30.17±0.35	40.75±0.50

<sup>1</sup>TEAC, <sup>2</sup>mg gallic acid equivalent/g extract, <sup>3</sup>PSD - Pooled Standard Deviation.



**FIGURE 1.** Sensory evaluation results for the three different samples of beverage. Different superscripts values indicate significant differences between them (ANOVA  $P \le 0.05$ ) in the attributes for aroma: 1000 mg kg<sup>-1 a,b</sup> – 2500 mg/kg<sup>a</sup> – 5000 mg/kg<sup>b</sup> flavor; 1000 mg/kg<sup>a</sup> – 2500 mg/kg<sup>a</sup> – 5000 mg/kg<sup>a</sup> – 5000 mg/kg<sup>a</sup> – 5000 mg/kg<sup>a</sup> – 5000 mg/kg<sup>b</sup>.



**FIGURE 2.** Preference test results for the three different samples of beverage. All samples have significant differences between them ( $P \le 0.05$ ).

## Conclusions

Solvent extraction with ethanol-water mixtures (up to 80% v/v ethanol) and higher solute-solvent ratios allow the extraction of compounds that have antioxidant capacity: phenolic compounds, such as flavonols and phenolic acids, and glucosinolates and carotenoids. This study showed a

promising application of nasturtium leaf extract in food products, more specifically in nectar-type drinks, with good reception by potential consumers.

#### Acknowledgments

This study was part of the research project "Desarrollo de un producto innovador a partir de la planta *Tropaeolum majus*" financed by MinCiencias, code 120380863525 and contract number 828 of 2018. The executing entity was the Pontificia Universidad Javeriana (Bogotá, Colombia).

#### **Conflict of interest statement**

The authors declare that there is no conflict of interests regarding the publication of this article.

#### Author's contributions

All authors conceptualized the study, CAAO developed the methodology, conducted the research, and wrote the manuscript. MSH administered the project, supervised the whole research, and reviewed the manuscript. AGT supervised the research. All authors reviewed the final version of the manuscript.

# Literature cited

- Alcalá Pedrajas, J. N. (2002). Intoxicación por metanol. Anales de Medicina Interna, 19(9), 494–495. https://doi.org/10.4321/ S0212-71992002000900020
- Alonso, J., & Desmarchelier, C. J. (2015). Capuchina. In J. Alonso, & C. J. Desmarcheller (Eds.), Plantas medicinales autóctonas de la Argentina: bases científicas para su aplicación en atención primaria de la salud (pp. 133–138). Corpus.
- Amiri, H. (2012). Volatile constituents and antioxidant activity of flowers, stems and leaves of *Nasturtium officinale* R. Br. *Natural Product Research*, 26(2), 109–115. https://doi.org/10.1 080/14786419.2010.534998
- Arellano, K., Herrera, J., Quispe, M., Espinosa, C., Veliz, N., & Orihuela, W. (2015). Evaluación de los compuestos fenólicos y capacidad antioxidante de tres colores de pétalos de mastuerzo (*Tropaeolum majus* L.). *Revista de la Sociedad Química del Perú*, 81(4), 319-327.
- Brondani, J. C., Cuelho, C. H. F., Marangoni, L. D., de Lima, R., Guex, C. G., Bonilha, I.F., & Manfron, M. P. (2016). Traditional usages, botany, phytochemistry, biological activity and toxicology of *Tropaeolum majus* L. - A review. *Boletín Latinoamericano y del Caribe de Plantas Medicinales y Aromaticas*, 15(4), 264–273.
- Celant, V. M., Braga, G. C., Vorpagel, J. A., & Salibe, A. B. (2016). Phenolic composition and antioxidant capacity of aqueous and ethanolic extracts of blackberries. *Revista Brasileira de Fruticultura*, 38(2), 1–8. https://doi.org/10.1590/0100-29452016411
- Demasi, S., Caser, M., Donno, D., Enri, S. R., Lonati, M., & Scariot, V. (2021). Exploring wild edible flowers as a source of bioactive compounds: New perspectives in horticulture. *Folia Horticulturae*, 33(1), 27–48. https://doi.org/10.2478/fhort-2021-0004
- Fernandes, L., Casal, S., Pereira, J. A., Saraiva, J., & Ramalhosa, E. (2017). Edible flowers: A review of the nutritional, antioxidant, antimicrobial properties and effects on human health. *Journal of Food Composition and Analysis*, 60, 38–50. https://doi. org/10.1016/j.jfca.2017.03.017
- Garzón, G. A., Manns, D., Riedl, K., Schwartz, J., & Padilla-Zakour, O. (2015). Identification of phenolic compounds in petals of nasturtium flowers (*Tropaeolum majus*) by high-performance liquid chromatography coupled to mass spectrometry and determination of oxygen radical absorbance capacity (ORAC). *Journal of Agricultural and Food Chemistry*, 63(6), 1803–1811. https://doi.org/10.1021/jf503366c
- Garzón, G. A., & Wrolstad, R. E. (2009). Major anthocyanins and antioxidant activity of Nasturtium flowers (*Tropaeolum*

*majus*). *Food Chemistry*, *114*(1), 44–49. https://doi.org/10.1016/j. foodchem.2008.09.013

- Hegnauer, R. (1973). Tropaeolaceae. Chemotaxonomie Der Pflanzen, 82(1931), 538–539. https://doi.org/10.1007/978-3-0348-9379-4\_49
- Hernández-Rodríguez, S., Quiroz-Reyes, C. N., Ramírez-Ortiz, M. E., Ronquillo-de Jesús, E., & Aguilar-Méndez, M. Á. (2020). Optimización del proceso de extracción asistida por ultrasonido de compuestos fenólicos de *Justicia spicigera* Schltdl. mediante la metodología de superficie de respuesta. *TIP Revista Especializada en Ciencias Químico-Biológicas*, 23, 1–7. https:// doi.org/10.22201/fesz.23958723e.2020.0.246
- Jakubczyk, K., Janda, K., Watychowicz, K., Łukasiak, J., & Wolska, J. (2018). Garden nasturtium (*Tropaeolum majus* L.) - a source of mineral elements and bioactive compounds. *Roczniki Panstwowego Zakladu Higieny*, 69(2), 119–126.
- Jha Kumar, A., & Sit, N. (2022). Extraction of bioactive compounds from plant materials using combination of various novel methods: A review. *Trends in Food Science and Technology*, 119, 579–591. https://doi.org/10.1016/j.tifs.2021.11.019
- Martínez-Navarrete, N., Camacho Vidal, M. M., & Martínez Lahuerta, J. J. (2008). Los compuestos bioactivos de las frutas y sus efectos en la salud. *Actividad Dietética*, *12*(2), 64–68. https://doi.org/10.1016/S1138-0322(08)75623-2
- Martínez-Patiño, J. C., Gullón, B., Romero, I., Ruiz, E., Brnčić, M., Žlabur, J. Š., & Castro, E. (2019). Optimization of ultrasoundassisted extraction of biomass from olive trees using response surface methodology. *Ultrasonics Sonochemistry*, *51*, 487–495. https://doi.org/10.1016/j.ultsonch.2018.05.031
- Medina-Medrano, J. R., Torres-Contreras, J. E., Valiente-Banuet, J. I., Mares-Quiñones, M. D., Vázquez-Sánchez, M., & Álvarez-Bernal, D. (2018). Effect of the solid-liquid extraction solvent on the phenolic content and antioxidant activity of three species of *Stevia* leaves. *Separation Science and Technology*, 54(14), 2283–2293. https://doi.org/10.1080/01496395.2018.1546741
- Navarro-González, I., González-Barrio, R., García-Valverde, V., Bautista-Ortín, A. B., & Periago, M. J. (2015). Nutritional composition and antioxidant capacity in edible flowers: Characterisation of phenolic compounds by HPLC-DAD-ESI/MS<sup>n</sup>. *International Journal of Molecular Sciences*, 16(1), 805–822. https://doi.org/10.3390/ijms16010805
- Yang, H., & Li, Q. (2022). Optimization of extraction process and the antioxidant activity spectrum–effect relationship of *Angelica dahurica*. *Biomedical Chromatography*, 36(4), Article e5322. https://doi.org/10.1002/bmc.5322

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