Study of the pollution by surfactants in a river of a high Andean micro basin

David Choque-Quispe [a], Mery Luz Masco-Arriola [b], Betsy Suri Ramos-Pacheco [a], Carlos Alberto Ligarda-Samanez [a], Aydeé Marilú Solano-Reynoso [c], Yudith Choque-Quispe [c] & Jhunior Felix Alonzo-Lanado [d]

[a] Departamento de Ingeniería y Tecnología Agroindustrial, Universidad Nacional José María Arguedas, Andahuaylas, Perú. dchoque@unajma.edu.pe, caligarda@unajma.edu.pe, bsramos@unajma.edu.pe, jhuniorfelix28@gmail.com
[b] Departamento de Ingeniería Química, Universidad Nacional de San Antonio Abad del Cusco, Perú. mery.masco@unsaac.edu.pe
[c] Escuela Profesional de Ingeniería Ambiental, Universidad Tecnológica de los Andes, Perú. ayma_21@hotmail.com, yuditchoque@gmail.com
[d] Departamento de Ingeniería Ambiental, Universidad Nacional José María Arguedas, Andahuaylas, Perú. jhuniorfelix28@gmail.com

Abstract
The surfactants linear alkylbenzene sulfonate (LAS), is used in large quantities in modern society, and wastewater containing it can contaminate groundwaters through runoff and sewers. The study aimed to determine the level of LAS in the Huatanay river; in Cusco city, Peru. Water samples were taken from nine points along the river, in the rainy and dry season in 2018. LAS surfactants were determined spectrometrically using methylene blue as indicator; dissolved oxygen (DO) and conductivity were also determined; both were correlated through Pearson coefficient at 5% significance, and analyzed in triplicate. The concentration of LAS in the dry season varied between 0.01 to 23.17 mg/L and in rains from 0.09 to 1.47 mg/L; the DO level shows values between 0.11 to 5.04 mg O_2/L, and the conductivity varied from 777.7 to 2688.9 µS/cm. In conclusion, high levels of LAS considerably decrease the DO and increase the conductivity

Keywords: conductivity; dissolved oxygen; high Andean micro basin; surfactants.

Estudio de la contaminación por surfactantes en un río de una microcuenca alto andina

Resumen
Los surfactantes lineales alquilbenceno sulfonato (LAS) se usa en grandes cantidades en la sociedad moderna, y las aguas residuales que lo contienen pueden contaminar las aguas subterráneas a través de la escorrentía y las alcantarillas. El estudio tuvo como objetivo determinar el nivel de LAS en el río Huatanay; en la ciudad del Cusco. Se tomaron muestras de agua de nueve puntos a lo largo del río, en temporada de lluvias y estiaje en el año 2017. Los surfactantes LAS se determinaron espectrométricamente utilizando azul de metileno como indicador; también se determinó el oxígeno disuelto (OD) y la conductividad; ambos se correlacionaron mediante el coeficiente de Pearson con significancia del 5%, y se analizaron por triplicado. La concentración de LAS en temporada de estiaje varió entre 0.01 a 23.17 mg/L y en lluvias de 0.09 a 1.47 mg/L; el nivel de OD muestra valores entre 0.11 a 5.04 mg O_2/L, y la conductividad varío de 777.7 a 2688.9 µS/cm. En conclusión, los altos niveles de LAS disminuyen considerablemente el OD y aumentan la conductividad.

Palabras clave: conductividad; microcuenca altoandina; oxígeno disuelto; surfactante.

1- Introduction

Emerging organic chemical compounds are resistant and have been shown to cause harm to the environment and human health [1,2], and the potential for toxicity when present in wastewater is still unknown [3], therefore, no acceptable limits have been established for those compounds [4], and furthermore, their exact chemical interaction is unknown [5].

Surfactants are widely used in household and industrial

products, which are discharged like wastewater after use [6],[7]. This account for further investigation about destination and impact of surfactants in the environment after release in sewage [3].

Due to their wide applicability, profitability, and level of consumption many commercial detergents contain anionic surfactants, including linear alkylbenzene sulphonates (LAS), being this the most important [5]. In general, commercial products contain between 25 to 30% of LAS, either as wetting agents, emulsifiers, foaming agents, in agricultural products, and pigments or paints [8].

It has been verified that LAS is aerobically biodegradable in wastewater, and its use is considered environmentally friendly [9]. According to Asok et al. [10], more than 90% are rapid biodegraded, approximately within 4 days. In addition, it was observed that biodegradation in absence of oxygen is not favorable [3].

LAS can cause harm in aquatic life and it is important to determine the concentrations in water to estimate the possible danger to aquatic organisms and living organisms [11]. Bioaccumulation creates a risk for organisms beyond the food chain due to the ability to absorb organic compounds [1], according to Ying [12] and Montagner et al. [13], concentrations higher than 0.1 mg/L of alkylphenols stimulate vitellogenin production in male fish, and its behavior in high Andean areas, above 3000 m altitude is not known.

Rivers and lagoons near urban areas receive discharges of wastewater, where there are not sewage treatment plants to reduce LAS residues, making the ecosystem a vulnerable place. So, the aimed of this study was to determine the level of LAS surfactants in the high Andean micro basin of the Huatanay river; in Cusco city, Peru at 3400 m of altitude.

2. Materials and methods

The study was carried out between January to July 2018. The water samples were taken along the Huatanay micro-basin river that belongs to the Vilcanota basin in Cusco (13° 31′ 06″ S, 71° 58′ 41″ W, and 3400 m of altitude). According to Köppen climate classification, the micro-basin has a Subtropical highland climate (Cwb), It presents well defined seasons, a wet season, with intense rainfall between October to March (from 500 to 1000 mm/year) and temperatures from 5 to 23°C; the dry season between April to September with temperature from 4 to 22°C. It presents an average relative humidity of 55%. Fig. 1 shows the distribution of rainfall during the experimental period.
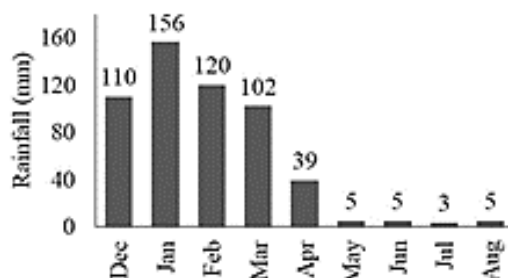

Figure 1. Rainfall from December to August 2018
Source: The Authors.

Table 1.
Description of the sampling points in the Huatanay river

Sampling points	Altitude (m)	Coordinates		Zone/Activity
		S	W	
P1	3376	13° 32′ 32′′	71° 58′ 56′′	Urban with cropping areas
P2	3345	13° 32′ 01′′	71° 58′ 09′′	Urban
P3	3318	13° 32′ 13′′	71° 57′ 31′′	Urban
P4	3258	13° 32′ 04′′	71° 55′ 09′′	Urban
P5	3251	13° 31′ 57′′	71° 54′ 52′′	Urban
P6	3245	13° 32′ 04′′	71° 54′ 43′′	Urban
P7	3216	13° 32′ 54′′	71° 53′ 28′′	Urban
P8	3199	13° 33′ 13′′	71° 52′ 23′′	Urban with cropping areas
P9	3186	13° 33′ 12′′	71° 51′ 45′′	Urban with cropping areas

Source: Authors

Nine points were selected along the river covering the urban area as shown in Table 1. The points vary from 3376 to 3186 m of altitude, and they are located along of 15.3 km and with a slope of 1.24%. The collecting samples were carried out in the afternoon from 14:00 at 18:00 hours and considering a day without rain.

Water samples were stored in amber glass bottles. The bottles were treated previously with nitric acid at 3% and rinse with ultrapure water type I. Samples were immediately refrigerated on ice and stored in dark until their analysis.

The concentration of LAS was determined at the AGP Laboratory - Labs in Lima, Peru, through a UV-Vis spectrophotometer at 652 nm, according to the 5540-C methodology - Anionic Surfactants as MBAS assay with the methylene blue colorimetric indicator [14]. On the other hand, the DO and the conductivity were determined in the field using a multiparameter following the 4500-O G and 2520-B methodology for their calibration and reading respectively [14]. The reading was carried out in triplicate at each sampling point, introducing the electrode in the river water with low turbulence and then, it left it to rest for 30 seconds until stabilization.

The variability of the results was evaluated through an analysis of variance (ANOVA) at 5% of significance, a Tukey means test ($p < 0.05$), and the correlation through the classified Pearson coefficient. Data analysis was performed with Minitab V17 software.

3. Results and discussion

Table 2, shows the results of the LAS surfactant concentration for both seasons at the sampling points. In rains season, the point P9 reported the higher concentration (1.47 mg/L). On the other hand, the concentration of LAS increases downstream from an initial value of 0.09 to 1.47 mg/L, showing a significant difference between the sampling points ($p < 0.05$).

This happens because the river runs through the downtown and many household wastewaters are discharged into the Huatanay river; although the LAS can be diluted during rain episodes. The point P9, in particular, receives the discharge from Cusco water treatment justifying the high concentration.

In the dry season, the concentration of LAS increased considerably due to the absence of dilution by rain. However, the point P1 presents quite lower concentration (0.01 mg/L). This zone is characterized with low population density and cropping areas. As the river flows downstream, the concentration reached to 23.17 mg/L (P3). In contrast to point P1, point P3 presents characteristics of household sewage collectors and car wash areas.

Table 2.
LAS concentration, DO and conductivity at the sampling points by season

Points	LAS (mg/L)		DO (mg O_2/L)		Conductivity (µS/cm)	
	Rains	Dry	Rains	Dry	Rains	Dry
P1	0.09 d ± 0.00	0.01 g ± 0.00	4.68 a ± 0.12	5.02 a ± 0.18	777.7 e ± 19.60	789.4 e ± 27.60
P2	0.10 d ± 0.00	10.27 c ± 0.36	4.16 b ± 0.10	1.4 d ± 0.05	847.8 e ± 21.40	806.2 e ± 28.20
P3	0.09 d ± 0.00	23.17 a ± 0.81	3.79 c ± 0.10	1.76 c ± 0.06	991.4 d ± 25.00	950.7 d ± 33.30
P4	0.03 e ± 0.10	8.77 d ± 0.31	4.14 b ± 0.10	5.04 a ± 0.18	2591.3 a ± 65.40	2688.9 a ± 94.10
P5	0.10 d ± 0.00	17.05 b ± 0.60	3.49 d ± 0.09	0.11 f ± 0.00	1370.4 b ± 34.60	1346.5 b ± 47.10
P6	0.22 b ± 0.01	4.18 f ± 0.15	4.15 b ± 0.10	2.35 b ± 0.08	1300.7 b ± 32.80	1296.3 b,c ± 45.40
P7	0.15 c ± 0.00	9.19 c,d ± 0.32	4.14 b ± 0.10	2.36 b ± 0.08	1378.4 b ± 34.80	1325.4 b,c ± 46.40
P8	0.16 c ± 0.00	6.51 e ± 0.23	3.49 d ± 0.09	0.27 f ± 0.01	1234.9 c ± 31.20	1206.0 c ± 42.20
P9	1.47 a ± 0.04	6.07 e ± 0.21	3.63 c,d ± 0.09	1.05 e ± 0.04	1013.6 d ± 25.60	992.3 d ± 34.70
	p < 0.05	p < 0.05	p < 0.05	p < 0.05	p < 0.05	p < 0.05

Means values with different letters are statistically significant at 5%.
Source: Authors

Table 3.
Pearson Correlation (Rs) for LAS, DO and conductivity

	Dry season		Rains season	
	LAS (mg/L)	DO (mg/L)	LAS (mg/L)	DO (mg/L)
DO (mg/L)	-0.30		-0.33	
Conductivity (µS/cm)	0.03	-0.38	0.20	0.03

Source: Authors

The waters from the Huatanay river are intensively collected in point 9 to use for irrigation purposes especially in dry season, at this point the concentration was 6.07 mg/L, which exceeds the Water Quality Standards – Peru (WQS) that recommend a maximum of 0.2 mg/L for this type of activity (MINAM, 2017).

Jiang et al. [15] and Ying [12] mention that concentrations higher than 0.1 mg/L induce chronic toxicity and bioaccumulation in plants [1,4,16]; in the same way this sewage can cause harm or destruction to the root cell membrane, changes in membrane permeability, in fine structure and effects on physiological processes such as photosynthesis [7]. Although, LAS requires at least 4 days for completely degradation [10], twice of this time is necessary in soil under aerobic conditions [17].

Regarding dissolved oxygen, it is observed that the levels are between 4.68 to 3.49 mg O_2/L (p < 0.05) in rains season, showing a tendency to decrease as the water runs down (Table 2), while in the dry season, the values are between 5.02 to 0.11 mg O_2/L (p < 0.05).

At points, P3, P8, and P9 in the rains season the DO levels are below the WQS [18] that recommends values higher than 4.0 mg O_2/L, while in the dry season only points P1 and P4 reach values recommended by WQS. These fluctuations are also due to the turbulence that the river presents near the sampling points [19] because it presents a 1.24% slope.

On the other hand, the conductivity shows values that vary from 777.7 to 2591.3 µS/cm in the rains season, while in the dry season shows values from 789.4 to 2688.9 µS/cm (Table 2). Conductivity increases as the downstream river flows; this fact is because detergents contribute dissolved salts into the water [20,21].

The Huatanay river receives domestic wastewater that is composed mainly by organic matter which requires oxygen for biodegradation of LAS [22]. Therefore, the decrease in the DO level would not only be attributed to the presence of LAS in the water samples. Such observation is indicated through the low negative correlation that it presents with the DO (Rs = - 0.30) (Table 3).

These reported values in most cases are lower than the stipulated by WQS for the conductivity to use as irrigation water [18]. However, the concentration of LAS presents a low positive correlation with the conductivity (Table 3) for dry season (Rs = 0.03) and rains season (Rs = 0.20), this is why LAS slightly increases conductivity.

4. Conclusions

This work presented the behavior of LAS surfactant residues and its relationship with conductivity and dissolved oxygen, in a high Andean river that crosses the urban area of the city of Cusco, located above 3400 m of altitude.

High levels of LAS detergents have been found in the Huatanay river in the dry season, which exceeds the values of the water quality standards for irrigation water, of the Peruvian regulations; the DO levels at most sampling points are below the values of the water quality standards, while conductivity is acceptable for irrigation water; the conductivity and DO, show a low negative and positive correlation respectively, with the detergent level for both seasons; so that high Andean river of the Huatanay that crosses the urban area of the city of Cusco, is contaminated by surfactants.

References

[1] Stuart, M., Lapworth, D., Crane, E. and Hart, A., Review of risk from potential emerging contaminants in UK groundwater. Science of the Total Environment, 416, pp. 1-21, 2012. DOI: 10.1016/j.scitotenv.2011.11.072

[2] Pal, A., He, Y., Jekel, M., Reinhard, M. and Gin, K.Y., Emerging contaminants of public health significance as water quality indicator compounds in the urban water cycle. Environmental International, 71, pp. 46-62, 2014. DOI: 10.1016/j.envint.2014.05.025

[3] Furlong, E.T., Batt, A.L., Glassmeyer, S.T., Noriega, M.C., Kolpin, D.W., Mash, H. and Schenck, K.M., Nationwide reconnaissance of contaminants of emerging concern in source and treated drinking waters of the United States: Pharmaceuticals. Sci Total Environ., 579, pp. 1629-1642, 2017. DOI: 10.1016/j.scitotenv.2016.03.128

[4] Baldwin, A.K., Corsi, S.R., De Cicco, L.A., Lenaker, P.L., Lutz, M.A., Sullivan, D.J., and Richards, K.D., Organic contaminants in Great Lakes tributaries: prevalence and potential aquatic toxicity. Science of the Total Environment, 554, pp. 42-52, 2016. DOI: 10.1016/j.scitotenv.2016.02.137

[5] Murguía, M.C., Cabrera, M.I., Guastavino, J.F. and Grau, R.J., New oligomeric surfactants with multiple-ring spacers: synthesis and tensioactive properties. Colloids and Surfaces A: Physicochem. Eng. Aspects, 262(2), pp. 1-7, 2005. DOI: 10.1016/j.colsurfa.2005.03.018

[6] Rasmussen, D., Slothuus, T., Petersen, G.I. and Madsen, T., Aquatic risk assessment of alcohol ethoxylates, alcohol ethoxysulphates and linear alkylbenzene sulphonate used in household detergents. Tenside Surfactants Detergents, 48(5), pp. 383-389, 2011. DOI: 10.3139/113.110143

[7] Cowan-Ellsberry, C., Belanger, S., Dorn, P., Dyer, S., McAvoy, D., Sanderson, H., Versteeg, D., Ferrer, D. and Stanton, K., Environmental

safety of the use of major surfactant classes in North America. Environ Sci Technol. 44(17), pp. 1893-1993, 2014. DOI: 10.1080/10739149.2013.803777

[8] Myers, D., Surfactant Science and Technology, 3rd ed. Wiley. New Jersey, U.S.A, 2010.

[9] Ojo, O.A. and Oso, B.A., Biodegradation of synthetic detergents in wastewater. African Journal of Biotechnology, 8(6), pp. 1090-1109, 2009.

[10] Asok, A.K., Fathima, P.A. and Jisha, M.S., Biodegradation of Linear Alkylbenzene Sulfonate (LAS) by *Immobilized Pseudomonas* sp. Advances in Chemical Engineering and Science, 5(5), pp. 465-475, 2015. DOI: 10.4236/aces.2015.54048

[11] Cantarero, S., Prieto, C.A. and López, I., Occurrence of high-tonnage anionic surfactants in Spanish sewage sludge. Journal of Environmental Management, 95, pp. 149-153, 2012. DOI: 10.1016/j.jenvman.2011.05.027

[12] Ying, G., Fate, behavior and effects of surfactants and their degradation products in the environment, Environment International, 32(3), pp. 417-431, 2006. DOI: 10.1016/j.envint.2005.07.004

[13] Montagner, C.C., Sodré, F.F., Acayaba, R.D., Vidal, C., Campestrini, I., Locatelli, M.A., Pescara, I.C., Albuquerque, A.F., Umbuzeiro, G.A. and Jardim, W.A., Ten Years-snapshot of the occurrence of emerging contaminants in drinking, surface and ground waters and wastewaters from São Paulo State, Brazil. J. Braz. Chem. Soc., 30(3), pp. 614-632, 2019. DOI: 10.21577/0103-5053.20180232

[14] APHA - American Public Health Association, Standard methods for the examination of water and wastewater, 22nd Washington, USA, 2012.

[15] Jiang, J.J., Lee, C.L., Fang, M.D., Boyd, K.G. and Gibb, S.W., Source apportionment and risk assessment of emerging contaminants: an approach of pharmaco-signature in water systems, PLoS ONE, 10(4), pp. 1-21, 2015. DOI: 10.1371/journal.pone.0122813

[16] Gordon, A.K., Muller, W.J., Gysman, N., Marshall, S.J., Sparham, C.J., Connor, S.M. and Whelan, M.J., Effect of laundry activities on in-stream concentrations of linear alkylbenzene sulfonate in a small rural South African river. Science of the Total Environment, 407(15), pp. 4465-4471, 2009. DOI: 10.1016/j.scitotenv.2009.04.023

[17] González, M.M., Martín, J., Santos, J.L., Aparicio, I. and Alonso, E., Degradation and environmental risk of surfactants after the application of compost sludge to the soil. Waste Manag., 32(7), pp. 1324-1331, 2012. DOI: 10.1016/j.wasman.2012.02.023

[18] MINAM – Ministerio del Ambiente, Estándares de Calidad Ambiental (ECA), Decreto Supremo N° 004-2017-MINAM – Perú, 2017.

[19] Choque-Quispe, D., Ligarda-Samanez, C.A., Ramos-Pacheco, B.S., Solano-Reynoso, A.M. and Quispe-Quispe, Y., Caffeine and UV-Vis scanning and the water quality index in the high-Andean watershed of the Chumbao river, Andahuaylas, Apurímac, Perú. Tecnología Química, 39(3), pp. 619-637, 2019.

[20] López, M.A. and Espinoza, E., The contribution of the type of detergent to domestic laundry graywater composition and its effect on treatment performance. Water, 8(214), pp. 1-10, 2016. DOI: 10.3390/w8050214

[21] Goel, G. and Kaur, S., A study on chemical contamination of water due to household laundry detergents. Journal of Human Ecology, 38(1), pp. 65-69, 2012. DOI: 10.1080/09709274.2012.11906475

[22] Henning, P., Verge, C., Cassani, G., Jensen, J., Holmstrup, M., Schraepen, N., Jorgense, E., Gavor, Z. and Temara, A., Risk assessment of linear alkylbenzene sulphonates, LAS, in agricultural soil revisited: robust chronic toxicity tests for Folsomia candida (Collembola), Aporrectodea caliginosa (Oligochaeta) and Enchytraeus crypticus (Enchytraeidae). Chemosphere, 69(6), pp. 872-879, 2007. DOI: 10.1016/j.chemosphere.2007.06.090

D. Choque-Quispe, received a Dr. in environment and sustainable development at the Universidad Andina del Cusco, Peru, and he is an Eng. candidate in water resources and Environmental at the Universidad Federal de Paraná, Brazil. Therefore, is a MSc. in food science and technology at the Universidad Nacional de San Antonio Abad del Cusco, Peru. He is a full-time professor and researcher in bioactive compounds and water treatment with biopolymers in the Department of Agroindustrial Engineering and Technology at the Universidad Nacional José María Arguedas, Andahuaylas, Peru.
ORCID: 0000-0003-4002-7526

M.L. Masco-Arriola, is a Dr. in Science, Technology and Environment at Universidad Nacional de Puno, Peru, and MSc. in Food Science at University of Illinois, United States. She is a full-time professor, researcher and Dean in the Chemical Engineering Faculty at Universidad Nacional San Antonio Abad del Cusco Peru.
ORCID: 0000-0002-5156-6464

B.S. Ramos-Pacheco, is a Dr. candidate in environment and sustainable development at the Universidad Andina del Cusco, Peru. Received a MSc. in environmental engineering at the Universidad Nacional de San Cristóbal de Huamanga, Ayacucho, Perú. She is a full-time professor at the school of agroindustrial engineering at the Universidad Nacional José María Arguedas, Andahuaylas, Peru. Has interest in researches that including quality and pollution of water resources, bioactive compounds and natural polymers.
ORCID: 0000-0002-0286-0632

C.A. Ligarda-Samanez, is a Dr. candidate in environment and sustainable development at the Universidad Andina del Cusco, Peru. Received a MSc. in food technology at the Universidad Nacional Agraria La Molina Lima, Perú. Received a BSc. Eng in civil engineering. He is a full-time professor and researcher in bioactive compounds, emerging compounds and water treatment with biopolymers at department of agro-industrial engineering and technology at the Universidad Nacional José María Arguedas, Andahuaylas, Peru.
ORCID: 0000-0001-7519-8355

A.M. Solano-Reynoso, received a Dr. in environment and sustainable development at the Universidad Andina del Cusco, Peru. She is a full-time professor at the school of environmental engineering at the Universidad Tecnológica de los Andes. She participates in researches related to water treatment with biopolymers and water quality in lakes.
ORCID: 0000-0002-1835-2210

Y. Choque-Quispe, received a MSc. in civil engineering – water resources at the Universidad Nacional de San Antonio Abad del Cusco, Peru. She is a professor at the school of environmental engineering at the Universidad Tecnológica de los Andes. She participates in researches related to water treatment with biopolymers and quality waters.
ORCID: 0000-0002-3690-7267

J.F. Alonzo Lanado, He is a MSc. candidate in Civil Engineering-Geotechnics. He is a professor at the School of Environmental Engineering at the José María Arguedas National University, Andahuaylas, Peru.
ORCID: 0000-0001-7888-860X