

# PHYSICOCHEMICAL COMPARISON OF COMMERCIAL VS. EXTRACTED $\beta$ -GLUCANS AND STRUCTURAL CHARACTERIZATION AFTER ENZYMATIC PURIFICATION

## COMPARACIÓN FÍSICOQUÍMICA ENTRE $\beta$ -GLUCANOS COMERCIALES Y EXTRAÍDOS, Y CARACTERIZACIÓN ESTRUCTURAL DESPUÉS DE PURIFICACIÓN ENZIMÁTICA

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### ABSTRACT

**Background:**  $\beta$ -glucans (1-3: 1-4) are soluble fibers applied to foods due to their technological properties (water binding capacity, viscosity, emulsification and stabilization) and their beneficial effects on health. The functional properties of  $\beta$ -glucans can be lost during the extraction and purification processes. The high viscosity of  $\beta$ -glucans is related to a high molecular weight and its physiological properties in the intestine. Therefore, to characterize the fiber after its extraction and purification is fundamental to understand its possible applications in foods. **Objectives:** characterize  $\beta$ -glucans extracted (E $\beta$ G) and compare them with three commercial  $\beta$ -glucans (C $\beta$ G-A, C $\beta$ G-B and C $\beta$ G-C) to identify its possible applications in foods and to evaluate if enzymatic purification affects molecular and structurally the  $\beta$ -glucans. **Methods:** barley  $\beta$ -glucans were extracted (E $\beta$ G), characterized by chemical analyzes, rheological behavior, and color, and compared to three commercial  $\beta$ -glucans samples. Then, the extract was purified and its structural and molecular characteristics were calculated. **Results:** E $\beta$ G contained  $64.38 \pm 3.54\%$  of  $\beta$ -glucans, high starch contamination ( $12.70 \pm 1.73\%$ ), high content of calcium (8894 mg/kg), pseudoplastic behavior, and dark color ( $L^* = 52.77 \pm 0.7$ ). All commercial samples showed low starch contamination, lighter color, and Newtonian behavior. After purification starch and protein contamination decreased ( $0.85 \pm 0.46\%$  and  $5.50 \pm 0.12\%$  respectively), increased the content of  $\beta$ G ( $69.45 \pm 0.81\%$ ) and increased brightness ( $L^* = 92.60 \pm 1.70$ ). Purified  $\beta$ -glucans (P $\beta$ G) showed a molar weight of  $690 \pm 1.6$  kDa and species with degree polymerization 3 (DP3) to 11 (DP11) were identified on the structure. **Conclusions:** E $\beta$ G extracts before the purification presented a high viscosity and contamination. The enzymatic purification process was effective and allowed to maintain a high molar mass of P $\beta$ G and its distinctive molecular structures (species with DP3 and DP4). The commercial samples C $\beta$ G-A and C $\beta$ G-B showed a low content of  $\beta$ -glucans. Finally, C $\beta$ G-C presented the best physicochemical and rheological properties for its subsequent application in food.

**Keywords:** Soluble fibers, healthy food, pseudoplastic behavior, viscosity, molecular mass.

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## RESUMEN

**Antecedentes:** los  $\beta$ -glucanos (1-3: 1-4) son fibras solubles aplicadas a los alimentos debido a sus propiedades tecnológicas (capacidad de retención de agua, viscosidad, emulsificación y estabilización) y a sus efectos beneficiosos en la salud. Las propiedades funcionales de los  $\beta$ -glucanos pueden perderse durante los procesos de extracción y purificación. La alta viscosidad de los  $\beta$ -glucanos está relacionada con un alto peso molecular y con sus propiedades fisiológicas en el intestino. Por lo tanto, caracterizar la fibra después de su extracción y purificación es fundamental para comprender sus posibles aplicaciones en alimentos. **Objetivos:** caracterizar  $\beta$ -glucanos extraídos (E $\beta$ G) y compararlos con tres marcas comerciales (C $\beta$ G-A, C $\beta$ G-B y C $\beta$ G-C) para identificar su futura aplicación en alimentos y evaluar si la purificación enzimática afecta molecular y estructuralmente los  $\beta$ -glucanos. **Métodos:** se extrajeron  $\beta$ -glucanos de cebada (E $\beta$ G), caracterizados por análisis químicos, comportamiento reológico y color, y se compararon con tres muestras comerciales. Posteriormente, el extracto (E $\beta$ G) se purificó y se identificaron sus características estructurales y su peso molecular. **Resultados:** E $\beta$ G contenía  $64.38 \pm 3.54\%$  de  $\beta$ -glucanos, alta contaminación con almidón ( $12.70 \pm 1.73\%$ ), alto contenido de calcio ( $8894 \text{ mg / kg}$ ), comportamiento pseudoplástico y color oscuro ( $L^* = 52.77 \pm 0.7$ ). Todas las muestras comerciales mostraron una baja contaminación con almidón, color más claro y comportamiento newtoniano. Después de la purificación de E $\beta$ G, la contaminación con almidón y proteína disminuyó ( $0.85 \pm 0.46\%$  y  $5.50 \pm 0.12\%$ , respectivamente), aumentó el contenido de  $\beta$ G ( $69.45 \pm 0.81\%$ ) y aumentó su luminosidad ( $L^* = 92.60 \pm 1.70$ ). Los  $\beta$ -glucanos purificados (P $\beta$ G) mostraron un peso molar de  $690 \pm 1,6 \text{ kDa}$  y se identificaron en la estructura especies con grado de polimerización desde 3 (GP3) hasta 11 (GP11). **Conclusiones:** los E $\beta$ G antes de la purificación presentaron alta viscosidad y contaminación. El proceso de purificación enzimática fue efectivo y permitió mantener una alta masa molar de la fibra y sus estructuras moleculares características (especies con GP3 y GP4). Las muestras comerciales C $\beta$ G-A y C $\beta$ G-B mostraron un bajo contenido de  $\beta$ -glucanos. Finalmente, la C $\beta$ G-C presentó las mejores propiedades fisicoquímicas y reológicas para su posterior aplicación en alimentos.

**Palabras clave:** Fibras solubles, alimentos saludables, comportamiento pseudoplástico, viscosidad, peso molecular.

## INTRODUCTION

Mixed-linkage  $\beta$  (1-3) and  $\beta$  (1-4)  $\beta$ -glucans are soluble fibers composed of D-glucose monomers with several applications in health, food and pharmaceutical fields.  $\beta$ -glucans are mainly known for their health benefits such as maintaining normal blood LDL-cholesterol concentrations and reducing postprandial glycemic responses (1).

The use of  $\beta$ -glucans in food is based on their potential to act as a thickening agent (2), emulsion stabilizer (3) and fat substitute (4), qualities related to their physical and molecular properties (molar weight, viscosity, molar rate, among others).

In spite of the nutritional and functional benefits of  $\beta$ -glucans, their extraction and purification involve a complex process that requires special attention to improve their yield and keep up their functional properties (5). Wet extraction allows the enrichment of multiple highly concentrated nutrients and improves yields (6). However, solvents and chemicals used on wet extraction can potentially

cause loss of functional properties of  $\beta$ -glucans. Besides, wet extraction methods generally yield fiber contaminated with partially solubilized starch (7). This contamination in  $\beta$ -glucans extracts is responsible for viscosity alterations (8). Thus, it is recommended to use enzymes such as amylases and proteases to eliminate the starch and proteins that could be adhered to the fibers (6).

Differences in the physicochemical characteristics of  $\beta$ -glucans are attributed mainly to genetic and environmental factors, as well as to differences in extraction methodologies, purification, depolymerization, and analytical methodologies used to calculate their molecular mass (9). Thereby, molecular mass and viscosity of  $\beta$ -glucans are affected by their physical state in the grain, betaglucanase activity, processing and storage conditions (10).

The importance of chemical characterization of extracted  $\beta$ -glucans before their application in foods has been widely emphasized. In that sense, (10) mention that the food industry has always

shown a keen interest in physicochemical and functional properties of nutraceutical compounds, as this will help in the choice of a specific type of compound with particular characteristics. The aim of the present work was to characterize barley  $\beta$ -glucans extracted (E $\beta$ G) and to compare it with three commercial  $\beta$ -glucans in order to identify the best soluble fiber for future application in food. Subsequently, the extracts (E $\beta$ G) were purified (P $\beta$ G) and molecular and structurally assessed to identify what extent these processes affect the characteristics of  $\beta$ -glucans.

## MATERIALS AND METHODS

### Materials

Commercial dehulled barley was supplied by SL Alimentos Ltd. (Paraná, Brazil). The kit for the determination and purification of  $\beta$ -glucans was obtained from Megazyme International Ltda (Wicklow, Ireland). The  $\beta$ -Glucan test kit content lichenase enzyme (specific, endo-(1-3)(1-4)- $\beta$ -D-glucan 4-glucanohydrolase) to breakdown the link endo (1-3) (1-4); beta-glucosidase enzyme para breakdown of  $\beta$ -Gluco-oligosaccharides in D-glucose and GOPOD reagent to assay the D-glucose produced. Two commercial brands of  $\beta$ -glucan (C $\beta$ G: A and B) were supplied by Polycell Technologies (Minnesota, USA) and one sample (C $\beta$ G C) was supplied by Cargill Company (Minnesota, USA) Commercial  $\beta$ -glucans were selected in order to identify the best soluble fiber for future application in food. MES/TRIS buffer, and 2,5-dihydroxybenzoic acid (DHB) were bought from Sigma-Aldrich, Brasil Ltda (Sao Paulo, Brazil).

### Extraction of $\beta$ -glucans

Barley  $\beta$ -glucan was extracted in according (11) with some modifications (HCL was not used to avoid the  $\beta$ -glucan degradation during extraction). The barley grains were ground to 0.5 mm, mixed with distilled water (1:10), adjusted to pH 8 with  $\text{CaCO}_3$  (20% w/v) to precipitate the starch, and heated in a water bath (30 min at  $50 \pm 2^\circ\text{C}$ ). The solution was cooled to room temperature and centrifuged (4850 g/30 min at  $4^\circ\text{C}$ ). The supernatant was recovered and an equal volume of ethanol (99 %) was added for  $\beta$ -glucan precipitation. After 16 hours at  $4^\circ\text{C}$ , the solution was centrifuged (3700 g/10 min at  $4^\circ\text{C}$ ). The  $\beta$ -glucan

precipitate was recovered and washed with 99 % ethanol. Excess moisture was removed at room temperature for 2 hours. The extracted  $\beta$ -glucans (E $\beta$ G) were kept in a desiccator for 24 hours prior to grinding in an IKA mill (Model A11 Basic) and sifted to 0.5 mm.

### Composition analysis

Chemical characterization of all samples was performed according to the American Association of Cereal Chemists (12) methods of analysis. The analysis were: moisture (Method 44.01.01); ash (Method 08.01.01); protein by Kjeldahl, using 6.25 as conversion factor (Method 46.10.01); lipids by Soxhlet extraction with petroleum ether (Method 30.10.01); total (FT), insoluble (FI) dietary fibers by enzymatic gravimetric procedure using the kit assay supplied by Megazyme Internacional Ltda. (Wicklow, Ireland) (Method 32.07.01); total starch and  $\beta$ -glucan by enzymatic procedure using assay kit the Megazyme Internacional Ltda. (Wicklow, Ireland) (Methods 76.13.01 and 32.23.01, respectively). Reported values are the means of three replicates. The results were expressed in grams per 100 grams (dry basis).

### Mineral content

Mineral content was performed according to the AOAC Method 968.08 (13) for calcium (Ca), iron (Fe), magnesium (Mg), and zinc (Zn), by flame atomic absorption spectrometry (F-AAS). The atomic absorption spectrometer (Analyst 200) was equipped with deuterium arc, background corrector, and Echelle optical resolution system. Acetylene (99.7% purity) was used as fuel to heat the atomizer system and compressed air was used as support gas. All samples were previously calcined at  $550^\circ\text{C}$ . Ash were diluted with HCl 8 mol/L. Hollow cathode lamps for the analyzed elements were from Lumina.

Phosphorus (P) was determined by molecular spectrometry and the AOAC Method 965.17, using a visible UV spectrophotometer with tungsten lamp, set at 420 nm (Evolution 60S, Thermo Scientific, Madison, USA). After calcination and complexation as molybdovanadate phosphoric acid, phosphorus was determined by interpolation in calibration curves prepared with  $\text{KH}_2\text{PO}_4$  in the concentration range from 1 to 20 mg/L. Sodium and potassium were determined by flame atomic emission spectroscopy (F-AES) at 589.0 nm with a flame

photometer (Analyser Comércio e Indústria Ltda., São Paulo, Brazil), after calcination and ash dilution with nitric acid ( $\text{HNO}_3$ ) 4 mol/L. The results of the samples were interpolated in calibration curves that ranged from 1 to 10 mg/L. (AOAC Method 969.23). The results were expressed in mg per 1 kg.

### Water binding capacity (WBC)

The WBC of the samples was measured using the method described by (5) with some modifications. A volume of 20 mL of distilled water was added to a pre-weighed centrifuge tube containing 200 mg of  $\beta$ -glucans. The hydrated samples were placed on a shaker at 25°C for 3 hours and centrifuged (4850 g/30 min at 25°C). The supernatant was discarded, and the tubes with the precipitate were weighed again. The WBC of each sample was expressed as the weight of water held by 1.0 g of  $\beta$ -glucan sample and was quantified by Equation 1.

$$\text{WBC} = \frac{(\text{Tube weight} + \text{sediment}) - (\text{Tube weight} + \text{sample})}{\text{Sample weight}} \quad (\text{Equation 1})$$

Analyzes were performed in triplicate for each sample

### Determination of pH

The samples were pre-diluted in distilled water at 1.5% (w/v), placed at 80°C for an hour, cooled to room temperature for two hours, all with continuous mixing. The pH was determined, according to method 943.02 (14) using a previously calibrated pH meter (Q400A-Quimis, Ltda. São Paulo, Brazil).

### Rheological assessment

The rheological behavior of aqueous dispersions (1.5% w/v) of  $\beta$ -glucans was assessed following the methodology used by (11) with modifications. It was performed on a rotational rheometer (Brookfield Engineering Laboratories model RVDV-III Ultra, Stoughton, MA, USA) using concentric cylinders: spindle S21 (for E $\beta$ G) and ULA (for commercial brands samples). Duplicate analysis was carried out for each sample. Rheological responses were analyzed by the Ostwald Waale model or Power law by Equation 2:

$$\eta_a = K(\dot{\gamma})^{n-1} \quad (\text{Equation 2})$$

Where:  $\eta_a$  = Apparent viscosity ( $\text{mPa}\cdot\text{s}^{-1}$ ),  $K$  = Consistency index ( $\text{Pa}\cdot\text{s}^n$ ),  $\dot{\gamma}$  = Deformation rate ( $\text{s}^{-1}$ ),  $n$  = flow behavior index (dimensionless).

### Color

The color was determined using a Chroma Meter CR-400 colorimeter (Konica Minolta, Osaka, Japan) with CIE-Lab system ( $L^*$ ,  $a^*$ ,  $b^*$ ) using D65 illuminant and 10° viewing angle. Where  $L^*$  defines the brightness of the product with  $L^* = 100$  as white and  $L^* = 0$  as black;  $a^*$  = red (+) or green (-) direction;  $b^*$  = yellow (+) or blue (-) direction of a product. The saturation index  $C^*$ , indicates the intensity of a hue, was calculated by Equation 3: The reported values are the means of five replicates.

$$C^* = \sqrt{(a^*)^2 + (b^*)^2} \quad (\text{Equation 3})$$

Where:  $C^*$ : Saturation index;  $a^*$ : Red coloration;  $b^*$ : Yellow coloration

### Purification of extracts

E $\beta$ G sample was diluted with MES/TRIS buffer 0.05 M (pH 8.2) in the ratio 2.5% (w/v) and stirred for 15 min to guarantee complete hydration and dispersion without forming clumps. Thereafter, 50  $\mu\text{L}$  of thermostable  $\alpha$ -amylase (3000 U/mL) was added and incubated (35 min/95°C). Samples were cooled (60°C), then 100  $\mu\text{L}$  of protease (50 mg/mL) were added and incubated (60°C/30 min). Samples were cooled, pH adjusted to 4.5 (HCl 5%) and 200  $\mu\text{L}$  amyloglucosidase (140 U/mL) was added and incubated at 60°C/30 min. The extract was added to ethanol 99% (4:1, v/v), rested for 15 min, centrifuged (4850 g/4°C/20 min) and the precipitate was washed with ethanol (99%). The purified fraction of  $\beta$ -glucans (P $\beta$ G) was freeze-dried and saved for further analysis. Protein, starch,  $\beta$ -glucans, WBC and color analysis were performed on the P $\beta$ G with the same methodologies previously described.

### Determination of molar weight

The determination of average molecular weight ( $M_w$ ), radius of gyration ( $R_g$ ) and the second virial coefficient ( $A_2$ ) were performed in NaOH solution (0.5 mol/L) following the method described by (15). Static measurements were conducted using a Brookhaven instruments including a precision goniometer (BI-200SM), and a monochromatic laser light source (Laser

He-Ne,  $\lambda = 632.8$  nm), with power of 35 mW (Brookhaven Instruments, Holtsville, NY, USA). Light scattering measurements were carried out in the angular range of 40 - 150° with intervals of 5°, using decalin as an internal reference. The signal was digitized by a discriminator-amplifier and analyzed by a multiple digital correlator. Purified  $\beta$ -glucans samples (P $\beta$ G) were diluted in several concentrations (0.1 – to 0.8 mg/mL) to ensure complete hydration and dispersion without forming clumps. The solutions were filtered through 0.45 mm cellulose ether filters and placed into glass cells cylindrical vials with 10 mm in diameter. From the intensity of light scattered at different angles ( $\theta$ ), for different concentrations of polymer, the Zimm Plot diagram was constructed using the software supplied with the equipment. All the light scattering measurements were performed at 25°C. The refractive index increment,  $dn/dc$ , was determined as 0.146 mL/g for  $\beta$ -glucan in aqueous solution. Five replicates were performed at each dilution. The result was expressed in kDa.

### Structural characterization by MALDI-TOF

$\beta$ -glucan hydrolysates (H $\beta$ -G) were prepared following the methodology used by (16) with modifications. Solutions (0.2 to 0.5% w/w) of P $\beta$ G in water HPLC grade were prepared by incubation in a boiling water bath with continuous stirring until the sample was fully solubilized. After cooling to room temperature, the pH was adjusted to 6.5 followed by the addition of 1.35 mL of lichenase (150 U). The mixture was incubated at 50°C/1 h, and stirred every 20 min. The hydrolysate was heated in a boiling water bath to inactivate the enzyme and centrifuged at 4580 g for 20 min. The supernatant was collected and subjected to matrix assisted laser desorption ionization time of flight (MALDI-TOF) analysis. The  $\beta$ -glucan hydrolysates (H $\beta$ -G) were mixed with three different matrices (DHB, Super-DHB, and DHB + Isonanilin) in the ratio of 1:3 and dried for 10 min at room temperature. The nitrogen laser pulse was accelerated under 20 kV before entering the time-of-flight mass spectrometer. The spectra were acquired in  $m/z$  range from 400 to 20,000 Da with low mass gate set to  $m/z = 400$ , 65% laser intensity, 2000 laser shots accumulated (200 to 200, in linear positive mode, reflector). The MALDI-TOF spectra were recorded on a MALDI-TOF Autoflex III Smartbeam Bruker Daltonics and processed and

calculated by PolyTools software from Flex Analysis (Bruker Daltonik).

### Data processing and statistical analysis

Data were processed with Statistica software (version 7.0, Statsoft Inc., Tulsa, OK, USA). Analysis of variance (One-way ANOVA) was performed using the Tukey's test to verify the difference between the means to  $p < 0.05$ .

## RESULTS

### Chemical analysis, pH and WBC

Extracted (E $\beta$ G) and commercial C $\beta$ G (A, B, C)  $\beta$ -glucans presented significant differences ( $p < 0.05$ ) for all nutrients analyzed, except for lipids as showed in the Table 1.

**Table 1.** Composition analysis of  $\beta$ -glucans samples in dry basis.

Content (g/100g)	Samples			
	C $\beta$ G A	C $\beta$ G B	C $\beta$ G C	E $\beta$ G
Moisture	11.10 $\pm$ 0.05 <sup>a</sup>	11.11 $\pm$ 0.06 <sup>a</sup>	4.61 $\pm$ 0.09 <sup>b</sup>	12.77 $\pm$ 0.05 <sup>c</sup>
Total Starch	1.68 $\pm$ 0.21 <sup>a</sup>	2.32 $\pm$ 0.30 <sup>a</sup>	4.60 $\pm$ 0.57 <sup>a</sup>	12.70 $\pm$ 1.73 <sup>b</sup>
Protein	16.10 $\pm$ 0.69 <sup>a</sup>	17.46 $\pm$ 0.04 <sup>a</sup>	1.97 $\pm$ 0.62 <sup>b</sup>	8.66 $\pm$ 0.34 <sup>c</sup>
Lipids	1.31 $\pm$ 0.31 <sup>a</sup>	1.01 $\pm$ 0.25 <sup>a</sup>	0.46 $\pm$ 0.03 <sup>a</sup>	1.84 $\pm$ 0.79 <sup>a</sup>
Ash	2.32 $\pm$ 0.18 <sup>a</sup>	2.57 $\pm$ 0.05 <sup>a</sup>	3.58 $\pm$ 0.01 <sup>b</sup>	5.23 $\pm$ 0.35 <sup>c</sup>
Insoluble fiber	17.35 $\pm$ 1.75 <sup>a</sup>	15.67 $\pm$ 1.02 <sup>a</sup>	1.30 $\pm$ 0.40 <sup>b</sup>	6.47 $\pm$ 0.81 <sup>c</sup>
$\beta$ -glucans (SF)	27.99 $\pm$ 0.80 <sup>a</sup>	29.97 $\pm$ 1.59 <sup>a</sup>	79.26 $\pm$ 0.51 <sup>b</sup>	64.38 $\pm$ 3.54 <sup>c</sup>
WBC (g/g)	2.04 $\pm$ 0.11 <sup>a</sup>	2.45 $\pm$ 0.09 <sup>a</sup>	3.68 $\pm$ 0.13 <sup>b</sup>	5.55 $\pm$ 0.24 <sup>c</sup>
pH**	5.99 $\pm$ 0.07 <sup>a</sup>	5.76 $\pm$ 0.10 <sup>b</sup>	6.56 $\pm$ 0.08 <sup>c</sup>	7.30 $\pm$ 0.09 <sup>d</sup>

Means  $\pm$  standard deviation followed by equal letters in the same line does not have significant difference by the Tukey's test ( $p > 0.05$ ).

\*\* Dimensionless. WBC = water binding capacity; SF = soluble fiber.

Particularly, C $\beta$ G A and C $\beta$ G B presented lower starch, higher protein, higher insoluble fiber (IF), and lower  $\beta$ -glucans (SF) levels. The bigger value of IF in the two commercial samples C $\beta$ G A and C $\beta$ G B seemed to indicate a possible dry extraction process of  $\beta$ -G. Dry extraction sequentially separates the pericarp, aleurone and subaleurone layers and embryo by abrasive cleaning to obtain concentrated fractions of  $\beta$ G still attached to high levels of IF (7) and (17). In addition, protein results from these samples (C $\beta$ G A and C $\beta$ G B) suggest that purification of the extracts was not performed. Soluble fibers ( $\beta$ -glucans) appeared in a larger proportion in C $\beta$ G C and E $\beta$ G ( $p < 0.05$ ).

The WBC was higher in the E $\beta$ G sample than in the sample C $\beta$ G C, followed by C $\beta$ G B and C $\beta$ G A ( $p < 0.05$ ) (Table 1). As E $\beta$ G and C $\beta$ G C have higher  $\beta$ -glucan content, it was expected that the higher WBC due to their ability of  $\beta$ -glucans to bind to water.

### Mineral content

The results of mineral content are showed in the Table 2. All samples presented a significant difference. The most abundant mineral in the samples was phosphorus, followed by potassium, calcium and magnesium.

**Table 2.** Mineral content of  $\beta$ -glucans samples.

Minerals (mg/kg)	Samples			
	C $\beta$ G A	C $\beta$ G B	C $\beta$ G C	E $\beta$ G
P	10113 $\pm$ 19 <sup>a</sup>	3244.2 $\pm$ 2.4 <sup>b</sup>	11934 $\pm$ 54 <sup>c</sup>	4711.8 $\pm$ 14.1 <sup>d</sup>
K	5966.8 $\pm$ 422.7 <sup>a</sup>	3159.9 $\pm$ 44.5 <sup>b</sup>	1017.3 $\pm$ 24.1 <sup>c</sup>	1611.9 $\pm$ 10.4 <sup>c</sup>
Mg	1795.6 $\pm$ 131.9 <sup>a</sup>	681.68 $\pm$ 38.70 <sup>b</sup>	2436.9 $\pm$ 19.8 <sup>c</sup>	303.62 $\pm$ 16.28 <sup>d</sup>
Ca	303.51 $\pm$ 20.18 <sup>a</sup>	333.64 $\pm$ 32.61 <sup>a</sup>	4729.4 $\pm$ 732.3 <sup>b</sup>	8894.2 $\pm$ 686.0 <sup>c</sup>
Na	43.670 $\pm$ 0.021 <sup>a</sup>	2499.5 $\pm$ 132.7 <sup>b</sup>	497.62 $\pm$ 4.16 <sup>c</sup>	145.55 $\pm$ 17.92 <sup>a</sup>
Zn	13.460 $\pm$ 0.042 <sup>a</sup>	10.765 $\pm$ 0.403 <sup>b</sup>	39.975 $\pm$ 0.250 <sup>c</sup>	9.8138 $\pm$ 0.9839 <sup>b</sup>
Fe	16.570 $\pm$ 0.014 <sup>a</sup>	8.580 $\pm$ 0.0000 <sup>b</sup>	26.810 $\pm$ 0.282 <sup>c</sup>	11.534 $\pm$ 1.380 <sup>d</sup>

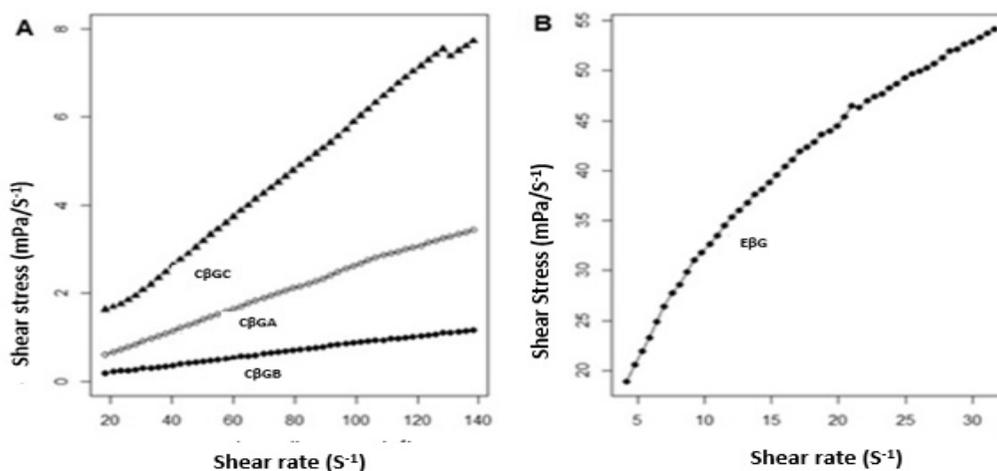
Means  $\pm$  standard deviation followed by equal letters in the same line does not have significant difference by the Tukey's test ( $p > 0.05$ ).

On the other hand, sodium concentration was significantly higher in the C $\beta$ G B samples than in the rest of the samples and the concentration of iron was higher in the C $\beta$ G C samples ( $p < 0.05$ ).

### Rheological assessment

In the commercial samples (C $\beta$ G-A, B, and C), the shear stress was proportional to the shear

rate, being a Newtonian behavior as shown in the Figure 1A. On the other hand, the sample E $\beta$ G presented a flow behavior similar to a pseudoplastic fluid where the viscosity decreases with the increase of the shear rate and the ratio between shear stress and shear rate is not linear as shown in the Figure 1B.



**Figure 1.** Flow-behavior of  $\beta$ -glucans (1.5%). A: Commercial samples. B: extracted barley  $\beta$ -glucans (E $\beta$ G).

The rheological behavior of all samples was well explained by the power law model ( $R^2 > 0.99$ ) as shown in the Table 3. The three commercial samples (C $\beta$ G-A, B and C) presented low consistency index (k) and flow behavior index (n) approximate to 1, keeping closer to the Newtonian behavior, exhibited proportionality between the shear stress and the shear rate.

**Table 3.** Power Law parameters ( $\eta = k (\dot{\gamma})^{(n-1)}$ ) of  $\beta$ -glucan samples.

Sample	Consistency index (k)	Flow behavior index (n)	Apparent viscosity (mPa/s)**	R <sup>2</sup>
C $\beta$ G A	0.045 $\pm$ 0.00 <sup>a</sup>	0.882 $\pm$ 0.00 <sup>a</sup>	28.27 $\pm$ 2.3 <sup>a</sup>	0.999
C $\beta$ G B	0.013 $\pm$ 0.00 <sup>a</sup>	0.908 $\pm$ 0.02 <sup>a</sup>	9.25 $\pm$ 0.15 <sup>a</sup>	0.997
C $\beta$ G C	0.116 $\pm$ 0.03 <sup>a</sup>	0.855 $\pm$ 0.02 <sup>a</sup>	65.15 $\pm$ 15.27 <sup>a</sup>	0.988
E $\beta$ G	9.919 $\pm$ 0.42 <sup>b</sup>	0.500 $\pm$ 0.00 <sup>b</sup>	1401.68 $\pm$ 51.23 <sup>b</sup>	0.992

Means  $\pm$  standard deviation followed by equal letters in the same line does not have significant difference by the Tukey's test ( $p > 0.05$ ).

\*\*Viscosity available in shear rate = 50 s<sup>-1</sup>.

On the other hand, E $\beta$ G had a high consistency index (k) and a mean value of flow behavior index (0.5), indicating pseudoplastic characteristics (Table 3).

### Color

In the Table 4 is showed color results. The highest values of luminosity (L\*) were presented by the C $\beta$ G-A followed by the C $\beta$ G-C, C $\beta$ G-B, and finally by the E $\beta$ G, with significant differences.

**Table 4.** Color results of  $\beta$ -glucan samples.

Sample	L*	a*	b*	C*
C $\beta$ G A	86.96 $\pm$ 0.3 <sup>a</sup>	2.43 $\pm$ 0.1 <sup>a</sup>	10.58 $\pm$ 0.4 <sup>a</sup>	10.86 $\pm$ 0.4 <sup>a</sup>
C $\beta$ G B	84.87 $\pm$ 0.2 <sup>b</sup>	2.71 $\pm$ 0.1 <sup>a</sup>	13.79 $\pm$ 0.3 <sup>b</sup>	14.06 $\pm$ 0.3 <sup>b</sup>
C $\beta$ G C	85.89 $\pm$ 0.4 <sup>c</sup>	0.95 $\pm$ 0.0 <sup>b</sup>	8.02 $\pm$ 0.3 <sup>c</sup>	8.08 $\pm$ 0.3 <sup>c</sup>
E $\beta$ G	52.77 $\pm$ 0.7 <sup>d</sup>	9.86 $\pm$ 0.4 <sup>c</sup>	18.78 $\pm$ 0.3 <sup>d</sup>	21.21 $\pm$ 0.3 <sup>d</sup>

Means  $\pm$  standard deviation followed by equal letters in the same line does not have significant difference by the Tukey's test ( $p > 0.05$ ).

The E $\beta$ G presented a darker color (Table 4). All samples presented values of parameter a\* lower than 10 suggesting that red tonality is not present

in great proportion in the samples analyzed. The highest values of b\* were found in E $\beta$ G and C $\beta$ G-B, which indicate that those samples were slightly yellowish.

### Purified extracts – P $\beta$ G

#### Protein, total starch, $\beta$ -G content, and WBC in purified extracts (P $\beta$ G)

After purification, total starch decreased to 0.85  $\pm$  0.46% in P $\beta$ G. Additionally, P $\beta$ G samples also had lower protein (5.50  $\pm$  0.12) and high  $\beta$ G (69.45  $\pm$  0.81) values. The WBC for P $\beta$ G was 3.47  $\pm$  0.38 (that is, each gram of P $\beta$ G can retain up to 3.47 grams of water). The P $\beta$ G presented color parameters with increased brightness L\* (92.60  $\pm$  1.70), decreased saturation index C\* (4.81  $\pm$  0.18), and showed a\* (0.81  $\pm$  0.08) and b\* (4.75  $\pm$  0.19) values which overall resulted in lighter samples with better appearance.

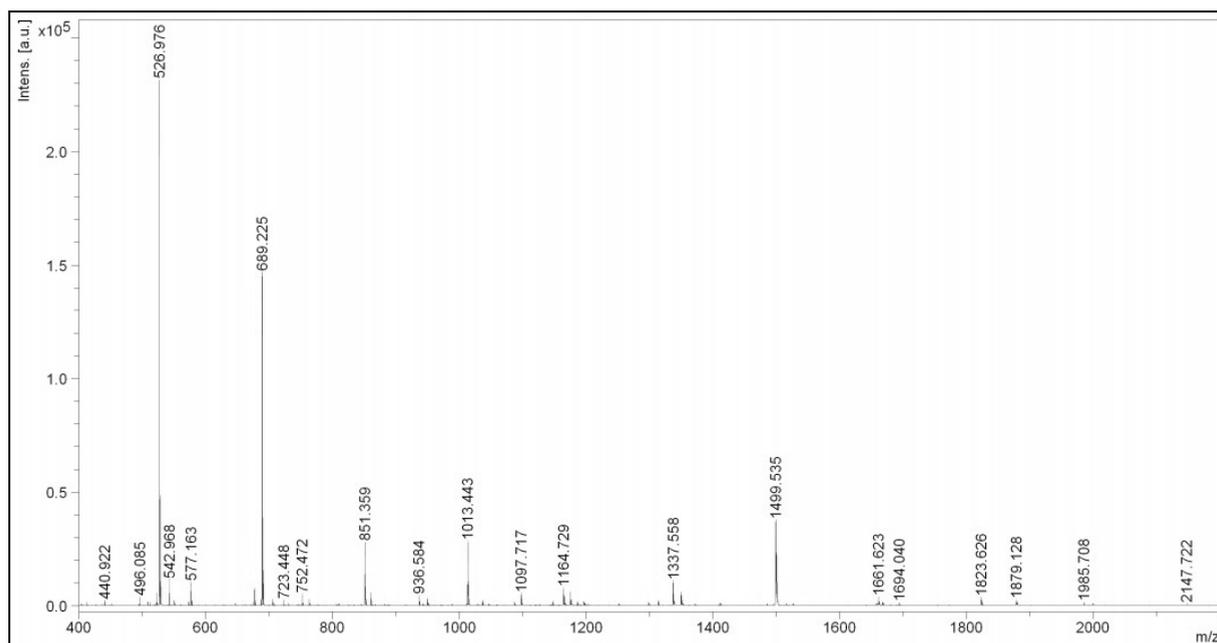
### Molar weight of P $\beta$ G

Since  $\beta$ -Glucans easily aggregate in aqueous solutions (because of inter-molecular hydrogen bonding), the Zimm Plot was performed in 0.5 mol/L NaOH to remove aggregates. The molar weight of P $\beta$ G by light scattering was 690  $\pm$  1.6 kDa with a gyration ( $R_g$ ) radius of 50  $\pm$  10 nm. The second virial coefficient  $A_2$  was found to be 2.30  $\pm$  0.37  $\times 10^3$  cm<sup>3</sup> mol/g<sup>2</sup>.

### Structural characteristics

P $\beta$ G was hydrolyzed with lichenase (H $\beta$ -G) and characterized by MALDI-TOF to identify affinity of H $\beta$ -G in three different matrices of ionization (DHB, Super-DHB, and DHB + Iovanilin) and obtain polysaccharides mass spectral profile. All matrices analyzed presented good MALDI-TOF signal by the clearest fragmentation pattern, however, DHB matrix (20 mg/mL 70% ACN/30% H<sub>2</sub>O/0.2% TFA) presented a high S/N ratio to fragments with DP > 5 as shown in the Figure 2.

In this study, the MALDI-TOF spectrum of hydrolyzed  $\beta$ -glucan (H $\beta$ -G) confirmed the presence of DP3 to DP11 species with DP3 (Mw = 526.976 Da) and DP4 (Mw = 689.976 Da) being the predominant species.



**Figure 2.** MALDI-TOF mass spectrum of barley  $\beta$ -glucan hydrolysate (H $\beta$ -G) in DHB matrix. Fragments DP3 (Mw: 526,976); Fragments DP4 (Mw: 689,225); Fragments DP5 (Mw: 851,359); Fragments DP6 (Mw: 1013,443); Fragments DP7 (Mw: 1164,729); Fragments DP8 (Mw: 1337,558); Fragments DP9 (Mw: 1499,535); Fragments DP10 (Mw: 1661,623); Fragments DP11 (Mw: 1823, 626).

## DISCUSSION

### Chemical analysis, pH and WBC

The E $\beta$ G sample presented a high starch level, indicative of larger contamination during fiber extraction. This happens in wet extraction, the starch is partially solubilized bonding to the  $\beta$ -glucan molecule (17). Overall, the main contaminants of  $\beta$ -glucans are starch and proteins present in the barley grain (18).  $\beta$ -glucan content is directly affected by extraction and purification processes used, therefore, interaction with other constituents of the barley grain is important in the technological perspectives of this polysaccharide (17). Barley  $\beta$ -glucan extraction under laboratory conditions resulted in similar values of starch 13.8%, protein 4.15% (11). Values of 15% of proteins were found in extracted oat  $\beta$ -glucans (19).

With regard to WBC, (20) presented similar values (3.54 - 4.77) in  $\beta$ -glucans gums. According to (21), the high WBC of the gums resulted in water absorption and intermolecular forces leading to the formation of aggregates or structures in the three-dimensional matrix. These interactions provided functional properties (gelling agents, thickeners, stabilizers, suspending agents) to the gums.

### Mineral content

The phosphorus content of commercial samples (C $\beta$ G-A, B and C) and the extracted (E $\beta$ G) found in this study are within the ranges reported by (19) and are higher than those reported by (5). The high content of calcium in the E $\beta$ G samples may be a consequence of the extraction because calcium carbonate was used during starch precipitation. The presence of other compounds such as protein, starch, metals and phosphorus can cause the cleavage and thermal degradation of  $\beta$ -glucans and consequently affect their molar mass and viscosity (19). According to (22), the presence of iron (Fe<sup>2+</sup>), can promote the formation of OH<sup>·</sup> radicals and consequently generate degradation of the polysaccharide, affecting its viscosity.

### Rheological assessment

Similar values of the rheological behavior were reported for extracts of barley  $\beta$ -glucans by (23), indicating values  $k = 0.06$  and  $n = 0.92$ . According to (18), pseudoplasticity of  $\beta$ -glucans has been established for high viscosity molecules and usually presents a high coefficient of consistency ( $k$ ) and low flow behavior index

(n). Viscosity values are directly proportional to WBC and pH. That is, when samples have higher pH, they have higher WBC and higher viscosity. Basic pH causes inactivation of the endogenous beta-glucanase preventing degradation of the fiber by the enzyme, keeping the molar mass of  $\beta$ -glucans and consequently their viscosity (24). Additionally, high pH values facilitate the solubility of the polysaccharides, mainly starch. This may facilitate the hydration of the starch and  $\beta$ -glucans present in the E $\beta$ G increasing its volume and consequently the viscosity of the solution. Nevertheless, the viscosity had no direct relationship with the  $\beta$ -glucan content in the samples, as expected. In the sample C $\beta$ G-C, the content of fiber soluble ( $\beta$ -glucan) was higher than E $\beta$ G sample, but the viscosity was lower. The high viscosity found in the E $\beta$ G sample can be explained by the presence of 12.7% of starch and 5.23% of ash. Additionally, concentration, solubility and molar mass, also affected the viscosity of  $\beta$ -glucan solutions (25). There is a direct relationship between viscosity increase and molar mass, reflected in the pseudo-elasticity characteristics of this compound and (9).

### Color

High  $L^*$  values (greater than 83) were related in  $\beta$ -glucans extracted by enzymatic methods, with low levels of starch and protein (5). However, (20) reported  $L^*$  values between 60.52 and 64.78 for  $\beta$ -glucans treated with high pressures at different dilutions. High degree of luminosity has an industrially appeal to add ingredients in other food products colors (5). Differences in the drying processes in C $\beta$ -G and E $\beta$ G can also influence the final hue. Thus, while C $\beta$ G is usually freeze-dried or spray-dried (17), the E $\beta$ G samples were air-dried and left in a desiccator, allowing the aggregation of the particles, oxidation reactions and probable enzymatic reactions leading to darkening.

### Protein, total starch, $\beta$ -G content, and WBC in purified extracts (P $\beta$ G)

The results of P $\beta$ G samples confirmed that WBC of  $\beta$ -glucans is maintained even after purification. WBC is an important physicochemical property of dietary fiber associated with viscosity

and their physiological effects (26). The WBC is attributed to the ability of the fiber to form networks that allow the retention of water through inter and intra molecular interactions.

### Molar weight of P $\beta$ G

The high value of  $M_w$  in the P $\beta$ G can be explained as consequence of the extraction and purification procedures that sought to maintain the integrity of the molecular structure. In this research, no was used protein precipitation procedure with HCl during the extraction. According to (24), pH values close to 4 can cause acid hydrolysis, decreasing the molar mass. The European Food Safety Authority (EFSA) indicated that the molar mass of barley  $\beta$ -glucans can range from 50 to 2000 kDa (1). In this sense, the molar mass of nine cultivars of Irish spring barley  $\beta$ -glucans were found in a range between 250.0 to 375.0 kDa using size exclusion chromatography (HPLC-SEC) (24) and the other hand, molar mass were found in a range between 150 to 581 kDa using dynamic light scattering (15). Differences in the molar mass of  $\beta$ -glucans have been attributed to extraction methodologies, purification, depolymerization events, and the analytical methodologies used to calculate molar mass (9).

### Structural characteristics of P $\beta$ G

DP3 and DP4 units are the main building blocks of barley  $\beta$ -glucans (16). According to (9), the major hydrolysates for cereal  $\beta$ -glucans are cellobiosyl-D-glucose (DP3) and cellotriosyl-D-glucose (DP4). However, others oligosaccharides are also released in smaller amounts from the polymer regions containing more than three consecutive linked glucose residues.

Our results indicated that starch and protein contaminations associated with extraction, as well as the drying process, affect the final quality of the E $\beta$ G. High viscosity in E $\beta$ G could difficult to use this fiber in concentrations greater than 1.5%. The use of enzymes is a highly effective method for purifying  $\beta$ -glucan samples improving their appearance without modifying their functional and structural characteristics

## CONCLUSIONS

The extraction conditions used in this study yielded extracts (E $\beta$ G) with high levels of fiber,

highly viscous, showed high WBC and dark color when compared to commercial samples, which makes it difficult to use this extract in concentrations of 1.5% to be included in foods.

The enzymatic purification process applied in the extract was effective removing contaminating starch and proteins, and improving the color while maintaining the WBC. Besides, the purified extract (P $\beta$ G) presented high molar mass and kept the main molecular structures of the fiber (DP3 and DP4 species). Preserve these characteristics in the soluble fiber ( $\beta$ -glucan) is important to its functional effect in the intestine when consumed. Therefore, the enzymatic purification of the E $\beta$ G can be recommended before its application in others foods.

On the other hand, two commercial  $\beta$ -glucans brands presented low fiber content (C $\beta$ G-A and C $\beta$ G-B). Consequently, the use of high amounts of these brands will be necessary to achieve values suggested for the consumption of soluble cereal fiber.

Finally, CBG-C presented higher purity, higher WBC, higher viscosity when compared to the other commercial samples. These samples at concentrations of 1.5% presented rheological properties that allow to be recommended for application in foods such as beverages, raw meat by injection or even incorporation in some foods, such as meat emulsions.

### CONFLICT OF INTEREST

The authors certify that there is no conflict of interest in this research.

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### AUTHORS' CONTRIBUTIONS

Authors made significant contributions to conception, design, conducting the experiments,

drafting, interpretation, and proofreading the manuscript. The authors participate in critical review of the article and give final approval of the version to be submitted.

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